

Classical Lifetime of a Bohr Atom

James D. Olsen and Kirk T. McDonald

Joseph Henry Laboratories, Princeton University, Princeton, NJ 08544

(March 7, 2005; updated May 30, 2017)

1 Problem

In the Bohr model [1] of the hydrogen atom's ground state, the electron moves in a circular orbit of radius $a_0 = 0.53 \times 10^{-10}$ m around the proton, which is assumed to be rigidly fixed in space. Since the electron is accelerating, a classical analysis suggests that it will continuously radiate energy, and therefore the radius of the orbit would shrink with time.

Considerations such as these in 1903 by J.J. Thomson [2, 3] led him to note that there is no radiation if the charge is distributed in space so as to form steady currents.¹ While a spinning shell or ring of charge provides a model for the magnetic moment of an atom, such charge configurations provide no restoring force against displacements of the center of the shell/ring from the nucleus. The (continuous) charge distribution must extend all the way to the nucleus if there is to be any possibility of classical electrostatic stability. A version of these insights was incorporated in Thomson's (not entirely self-consistent) model [2, 3] of the atom as a kind of "plum pudding" where the nucleus had a continuous, extended charge distribution in which more pointlike electrons were embedded. In this context, the measurements of Geiger, Marsden and Rutherford [5, 6, 7, 8, 9, 10] of α -particle scattering, which showed that the nucleus was compact, came as something of a surprise, and re-opened the door to models such as that of Bohr [1] in which pointlike electrons orbited pointlike nuclei, and radiation was suppressed by a "quantum" rule.

Note that the model of the atom that emerged following Schrödinger contains some classically agreeable features if (ground states of) atoms are not to radiate: The electron in an atom is considered to have a wavefunction that extends to the origin. The electric current associated with the electron is steady, and hence would not radiate if this were a classical current.

- a) Assuming that the electron is always in a nearly circular orbit and that the rate of radiation of energy is sufficiently well approximated by classical, nonrelativistic electrodynamics, how long is the **fall time** of the electron, *i.e.*, the time for the electron to spiral into the origin?
- b) The charge distribution of a proton has a radius of about 10^{-15} m, so the classical calculation would be modified once the radius of the electron's orbit is smaller than this. But even before this, modifications may be required due to relativistic effects.

Based on the analysis of part a), at what radius of the electron's orbit would its velocity be, say $0.1c$, where c is the speed of light, such that relativistic corrections become significant? What fraction of the electron's fall time remains according to part a) when the velocity of the electron reaches $0.1c$?

¹Several subsequent authors have claimed the existence of radiationless orbital motion of classical charges. However, these claims all appear to have defects. See, for example, [4].

- c) Do the relativistic corrections increase or decrease the fall time of the electron?

It suffices to determine the sign of the leading correction as the radial velocity of the radiating electron approaches the speed of light.

A question closely related to the present one is whether the rate of decay of the orbit of a binary pulsar system due to gravitational radiation is increased or decreased by special-relativistic “corrections” as the orbital velocity becomes relativistic.

2 Solution

- a) The dominant energy loss is from electric dipole radiation, which obeys the Larmor formula [11] (in Gaussian units),

$$\frac{dU}{dt} = -\langle P_{E1} \rangle = -\frac{2e^2 a^2}{3c^3}, \quad (1)$$

where a is the acceleration of the electron. For an electron of charge $-e$ and (rest) mass m_0 in an orbit of radius r about a fixed nucleus of charge $+e$, the radial component of the nonrelativistic force law, $\mathbf{F} = m_0 \mathbf{a}$, tells us that,

$$\frac{e^2}{r^2} = m_0 a_r \approx m_0 \frac{v_\theta^2}{r}, \quad (2)$$

in the adiabatic approximation that the orbit remains nearly circular at all times. In the same approximation, $a_\theta \ll a_r$, *i.e.*, $a \approx a_r$, and hence,

$$\frac{dU}{dt} = -\frac{2e^6}{3r^4 m_0^2 c^3} = -\frac{2r_0^3}{3r^4} m_0 c^3. \quad (3)$$

where $r_0 = e^2/m_0 c^2 = 2.8 \times 10^{-15}$ m is the classical electron radius. The total nonrelativistic energy (kinetic plus potential) is, using eq. (2),

$$U = -\frac{e^2}{r} + \frac{1}{2} m_0 v^2 = -\frac{e^2}{2r} = -\frac{r_0}{r} m_0 c^2. \quad (4)$$

Equating the time derivative of eq. (4) to eq. (3), we have,

$$\frac{dU}{dt} = \frac{r_0}{2r^2} \dot{r} m_0 c^2 = -\frac{2r_0^3}{3r^4} m_0 c^3, \quad (5)$$

or,

$$r^2 \dot{r} = \frac{1}{3} \frac{dr^3}{dt} = -\frac{4}{3} r_0^2 c. \quad (6)$$

Hence,

$$r^3 = a_0^3 - 4r_0^2 c t. \quad (7)$$

The time to fall to the origin is,

$$t_{\text{fall}} = \frac{a_0^3}{4r_0^2 c}. \quad (8)$$

With $r_0 = 2.8 \times 10^{-15}$ m and $a_0 = 5.3 \times 10^{-11}$ m, $t_{\text{fall}} = 1.6 \times 10^{-11}$ s.

This is of the order of magnitude of the lifetime of an excited hydrogen atom, whose ground state, however, appears to have infinite lifetime.

Since the kinetic energy of the electron goes to infinity as it spirals into the nucleus, the motion can be called a “runaway solution.”

b) The velocity v of the electron has components,

$$v_r = \dot{r} = -\frac{4}{3} \frac{r_0^2}{r^2} c, \quad (9)$$

using eq. (6), and,

$$v_\theta = r\dot{\theta} = \sqrt{\frac{e^2}{m_0 r}} = \sqrt{\frac{r_0}{r}} c, \quad (10)$$

according to eq. (2).

The azimuthal velocity is much larger than the radial velocity so long as $r \gg r_0$. Hence, $v/c \approx v_\theta/c$ equals 0.1 when $r_0/r \approx 0.01$, or $r \approx 100r_0$.

When $r = 100r_0$ the time t is given by eq. (7) as,

$$t = \frac{a_0^3 - r^3}{4r_0^2 c}, \quad (11)$$

so that,

$$\frac{t_{\text{fall}} - t}{t_{\text{fall}}} = \frac{r^3}{a_0^3} = \left(\frac{2.8 \times 10^{-13}}{5.3 \times 10^{-11}} \right)^3 \approx 1.5 \times 10^{-7}. \quad (12)$$

For completeness, we record other kinematic facts in the adiabatic approximation.

The angular velocity $\dot{\theta}$ follows from eq. (10) as,

$$\dot{\theta} = \sqrt{\frac{r_0}{r^3}} c. \quad (13)$$

The second time derivatives are thus,

$$\ddot{r} = \frac{8}{3} \frac{r_0^2}{r^3} \dot{r} c = -\frac{32}{9} \frac{r_0^4}{r^5} c^2, \quad \ddot{\theta} = -\frac{3}{2} \sqrt{\frac{r_0}{r^5}} \dot{r} c = 2 \sqrt{\frac{r_0}{r}} \frac{r_0^2}{r^4} c^2. \quad (14)$$

The components of the acceleration are,

$$a_r = \ddot{r} - r\dot{\theta}^2 = -\frac{32}{9} \frac{r_0^4}{r^5} c^2 - \frac{r_0}{r^2} c^2 \approx -\frac{r_0}{r^2} c^2 = -r\dot{\theta}^2, \quad (15)$$

$$a_\theta = 2\dot{r}\dot{\theta} + r\ddot{\theta} = -\frac{8}{3} \sqrt{\frac{r_0}{r}} \frac{r_0^2}{r^3} c^2 \ll a_r. \quad (16)$$

- c) We now examine the leading relativistic corrections to the nonrelativistic analysis of part a).

First, we recall that the lab frame rate of radiation by an accelerated charge obeys the Larmor formula (1) provided we use the acceleration in the instantaneous rest frame rather than in the lab frame. This is true because both dU and dt transform like the time components of a four-vector, so their ratio is invariant.

In the adiabatic approximation, the acceleration is transverse to the velocity. That is, $v_r \ll v_\theta$ from eqs. (9) and (10), while $a_r \gg a_\theta$ from eqs. (15) and (16). Therefore,

$$a^* = \gamma^2 a, \quad (17)$$

where a is the lab-frame acceleration, a^* is the acceleration in the instantaneous rest frame, and $\gamma = 1/\sqrt{1 - v^2/c^2} = 1/\sqrt{1 - \beta^2}$. Equation (17) holds because $a^* = d^2l^*/dt^{*2}$, and $dl^* = dl$ for motion transverse to the velocity of the electron, while the time-dilation is $dt^* = dt/\gamma$. Thus, the rate of radiation of energy by a relativistic orbiting electron is,

$$\frac{dU}{dt} = -\langle P_{E1} \rangle = -\frac{2e^2 a^{*2}}{3c^3} = -\frac{2\gamma^4 e^2 a_r^2}{3c^3}. \quad (18)$$

[If the acceleration were parallel to the velocity, $a^{**} = \gamma^3 a$ since now there would also be the Lorentz contraction, $dl = dl^*/\gamma$.]

The adiabatic orbit condition (2) for a relativistic electron becomes,

$$\frac{e^2}{r^2} = \gamma m_0 a_r = \gamma m_0 \frac{v_\theta^2}{r} \approx \gamma m_0 \frac{v^2}{r}. \quad (19)$$

This can be thought of as the transform of the rest-frame relation $eE_r^* = dP_r^*/dt^*$ upon noting that $E_r^* = \gamma E_r$ since the electric field is transverse to the velocity, $dt^* = dt/\gamma$, and $dP_r^* = dP_r = \gamma m_0 dv_r$.

Combining eq. (18) with the first form of eq. (19), we have,²

$$\frac{dU}{dt} = -\frac{2\gamma^2 e^6}{3m_0^2 c^3 r^4} = -\frac{2}{3} \gamma^2 \frac{r_0^3}{r^4} m_0 c^3. \quad (20)$$

We also rewrite eq. (19) as,

$$\frac{e^2}{m_0 c^2 r} = \frac{r_0}{r} = \gamma \frac{v^2}{c^2} = \gamma \beta^2 \approx \gamma \left(1 - \frac{1}{\gamma^2}\right), \quad (21)$$

and hence,

$$\gamma^2 - \gamma \frac{r_0}{r} - 1 = 0, \quad (22)$$

²The result of eq. (20) was perhaps first given by Heaviside [12], before the theory of relativity was developed. What we call “radiation,” Heaviside called “waste.”

$$\gamma = \frac{\frac{r_0}{r} + \sqrt{\frac{r_0^2}{r^2} + 4}}{2} = \sqrt{1 + \frac{r_0^2}{4r^2}} + \frac{r_0}{2r} \approx 1 + \frac{r_0}{2r} + \frac{r_0^2}{8r^2}. \quad (23)$$

The total lab-frame energy is now,

$$U = \gamma m_0 c^2 - \frac{e^2}{r} = \left(\gamma - \frac{r_0}{r} \right) m_0 c^2 \approx \left(1 - \frac{r_0}{2r} + \frac{r_0^2}{8r^2} \right) m_0 c^2, \quad (24)$$

using eq. (23). Then,

$$\begin{aligned} \frac{dU}{dt} &\approx \left(\frac{r_0}{2r^2} - \frac{r_0^2}{4r^3} \right) \dot{r} m_0 c^2 = \left(1 - \frac{r_0}{2r} \right) \frac{r_0}{2r^2} \dot{r} m_0 c^2 \\ &= -\frac{2}{3} \gamma^2 \frac{r_0^3}{r^4} m_0 c^3, \end{aligned} \quad (25)$$

using eq. (20). Finally,

$$\dot{r} \approx -\frac{4}{3} \gamma^2 \frac{r_0^2}{r^2} \frac{c}{1 - \frac{r_0}{2r}} \approx -\frac{4}{3} \frac{r_0^2}{r^2} c \left(1 + \frac{3r_0}{2r} \right), \quad (26)$$

which is larger than the nonrelativistic result (6) by the factor $1 + 3r_0/2r$. Hence, the fall time of the electron is **decreased** by the relativistic corrections.

We note that the relativistic corrections increased the rate of radiation, and decreased the factor A in the relation $dU/dt = A\dot{r}$. Hence, both of these corrections lead to an increase in the radial velocity \dot{r} , and to a decrease in the corresponding fall time of the electron.

2.1 Where Does the Radiated Energy Come From?

Comments by Larmor on p. 512 of [11] suggest that he considered the energy radiated by an accelerating charge to come from the kinetic energy of that charge. Similarly, in the last paragraph of [12] one reads “The kinetic energy of molecules is the natural source of the radiation....” And it can be that some people hold such views even today.

However, the present example illustrates what is called the “satellite paradox” [13, 14, 15] that if an object is subject to “friction” when in orbit in a $1/r$ potential, then the kinetic energy actually increases as a result. Since the radiation by a charge in a classical orbit about a fixed positive charge is a kind of “friction,” and the Coulomb potential goes as $1/r$, the velocity and the kinetic energy of the charge increase with time. Hence, it does not seem appropriate in this case to claim that the radiated energy came from the kinetic energy of the moving charge.

We can say that the radiation lowers the total energy U of the charge $-e$, which for its circular orbit about charge Ze , with potential $\phi = Ze/r$ is $U = -Ze^2/2r$ (since the kinetic energy is $-1/2$ of the potential energy $V = -Ze^2/r$). Hence, as U decreases due to the emission of radiation, r becomes smaller and the potential energy V decreases. We might then say that the radiated energy came from the potential energy.

In Maxwell's view, the potential energy V of charge $-e$ together with nucleus of charge Ze is the interaction energy of the electric field $\mathbf{E} = \mathbf{E}_{-e} + \mathbf{E}_{Ze}$ of the system,

$$V = U_{\text{int}} = \int \frac{\mathbf{E}_{-e} \cdot \mathbf{E}_{Ze}}{4\pi} d\text{Vol}. \quad (27)$$

So, our last view is that the radiated energy came from the interaction field energy.³

2.2 Effect of the Self Force/Radiation Reaction

In the preceding analysis we have ignored the self force $3e^3 \ddot{\mathbf{v}}/3c^3$ on the accelerated electron (for $v \ll c$, as first deduced by Lorentz [21] in a model of an electron of finite radius, and also found by Planck [22] who followed Poincaré [23] in supposing that there is reaction force on the electron associated with its emission of electromagnetic radiation. It has been argued [24] that if this self force/radiation reaction is included in the equations of motion of the electron, it does not prevent a “runaway” solution (although the classical lifetime is extended somewhat).

References

- [1] N. Bohr, *On the Constitution of Atoms and Molecules*, Phil. Mag. **26**, 1 (1913), http://kirkmcd.princeton.edu/examples/QM/bohr_pm_26_1_13.pdf
On the Constitution of Atoms and Molecules. PART II.—Systems Containing only a Single Nucleus, Phil. Mag. **26**, 476 (1913), http://kirkmcd.princeton.edu/examples/QM/bohr_pm_26_476_13.pdf
- [2] J.J. Thomson, *The Magnetic Properties of Systems of Corpuscles describing Circular Orbits*, Phil. Mag. **6**, 673 (1903), http://physics.princeton.edu/~mcdonald/examples/EM/thomson_pm_45_673_03.pdf
- [3] J.J. Thomson, *On the Structure of the Atom: an Investigation of the Stability and Periods of Oscillation of a number of Corpuscles arranged at equal intervals around the Circumference of a Circle; with Application of the results to the Theory of Atomic Structure. The Magnetic Properties of Systems of Corpuscles describing Circular Orbits*, Phil. Mag. **7**, 237 (1904), http://kirkmcd.princeton.edu/examples/EM/thomson_pm_7_237_04.pdf
- [4] P. Pearle, *Absence of Radiationless Motions of Relativistically Rigid Classical Electron*, Found. Phys. **7**, 931 (1977), http://physics.princeton.edu/~mcdonald/examples/EM/pearle_fp_7_931_77.pdf
- [5] H. Geiger, *On the Scattering of α -Particles by Matter*, Proc. Roy. Soc. London **81**, 174 (1908), http://kirkmcd.princeton.edu/examples/EP/geiger_prsla_81_174_08.pdf

³For other examples of the Maxwellian view the energy changes of an interacting charge can/should be related to changes in field energy, see [16]. This view is also relevant to the famous case of a uniformly accelerated charge [17], where the “radiation” does not come from the charge, but is a result of rearrangement of the field energy of the charge. See also the case of an exponentially decaying electric dipole [18, 19] and sec. 2.5 of [20].

- [6] H. Geiger and E. Marsden, *On a Diffuse Reflection the α -Particles*, Proc. Roy. Soc. London **82**, 495 (1909), http://kirkmcd.princeton.edu/examples/EP/geiger_prsla_82_495_09.pdf
- [7] H. Geiger, *The Scattering of α -Particles by Matter*, Proc. Roy. Soc. London **83**, 492 (1910), http://kirkmcd.princeton.edu/examples/EP/geiger_prsla_83_492_10.pdf
- [8] E. Rutherford, *The Scattering of α and β Particles by Matter and the Structure of the Atom*, Phil. Mag. **21**, 669 (1911),
http://kirkmcd.princeton.edu/examples/EP/rutherford_pm_21_669_11.pdf
- [9] H. Geiger and E. Marsden, *The Laws of Deflexion of α Particles through Large Angles*, Phil. Mag. **25**, 604 (1913), http://kirkmcd.princeton.edu/examples/EP/geiger_pm_25_604_13.pdf
- [10] E. Rutherford, *The Structure of the Atom*, Phil. Mag. **27**, 488 (1914),
http://kirkmcd.princeton.edu/examples/EP/rutherford_pm_27_488_14.pdf
- [11] See the last page of J. Larmor, *On the Theory of the Magnetic Influence on Spectra; and on the Radiation from moving Ions*, Phil. Mag. **44**, 503 (1897),
http://kirkmcd.princeton.edu/examples/EM/larmor_pm_44_503_97.pdf
- [12] O. Heaviside, *The Waste of Energy from a Moving Electron*, Nature **67**, 6 (1902),
http://kirkmcd.princeton.edu/examples/EM/heaviside_nature_67_6_02.pdf
- [13] B.D. Mills, Jr., *Satellite Paradox*, Am. J. Phys. **27**, 115 (1959),
http://kirkmcd.princeton.edu/examples/mechanics/mills_ajp_27_115_59.pdf
- [14] L. Blitzer, *Satellite Orbit Paradox: A General View*, Am. J. Phys. **39**, 882 (1971),
http://kirkmcd.princeton.edu/examples/mechanics/blitzer_ajp_39_882_71.pdf
- [15] F.P.J. Rimrott and F.A. Salustri, *Open Orbits in Satellite Dynamics*, Tech. Mech. **21**, 207 (2001), http://kirkmcd.princeton.edu/examples/mechanics/rimrott_tm_21_207_01.pdf
- [16] M.S. Zolotorev, S. Chattopadhyay and K.T. McDonald, *A Maxwellian Perspective on Particle Acceleration* (Feb. 24, 1998), <http://kirkmcd.princeton.edu/examples/vacuumaccel.pdf>
- [17] V. Onoichin and K.T. McDonald, *Fields of a Uniformly Accelerated Charge* (Aug. 19, 2014), <http://kirkmcd.princeton.edu/examples/schott.pdf>
- [18] L. Mandel, *Energy Flow from an Atomic Dipole in Classical Electrodynamics*, J. Opt. Soc. Am. **62**, 1011 (1972), http://kirkmcd.princeton.edu/examples/EM/mandel_josa_62_1011_72.pdf
- [19] H.G. Schantz, *The flow of electromagnetic energy in the decay of an electric dipole*, Am. J. Phys. **63**, 513 (1995), http://kirkmcd.princeton.edu/examples/EM/schantz_ajp_63_513_95.pdf
Electromagnetic Energy Around Hertzian Dipoles, IEEE Ant. Prop. Mag. **43**, 50 (2001),
http://kirkmcd.princeton.edu/examples/EM/schantz_ieeeapm_43_50_01.pdf
- [20] K.T. McDonald, *The Fields of a Pulsed, Small Dipole Antenna* (Mar. 16, 2007),
http://kirkmcd.princeton.edu/examples/pulsed_dipole.pdf

- [21] H.A. Lorentz, *La Théorie Électromagnétique de Maxwell et son Application aux Corps Mouvants*, Arch. Néerl. **25**, 363 (1892),
http://kirkmcd.princeton.edu/examples/EM/lorentz_theorie_electromagnetique_92.pdf
- [22] M. Planck, *Über electrische Schwingungen, welche durch Resonanz erregt und durch Strahlung gedämpft werden*, Ann. d. Phys. **60**, 577 (1897),
http://kirkmcd.princeton.edu/examples/EM/planck_ap_60_577_97.pdf
- [23] H. Poincaré, *Sur la théorie des oscillations hertziennes*, Compte Rendus Acad. Sci. **113**, 515 (1891), http://kirkmcd.princeton.edu/examples/EM/poincare_cras_113_515_91.pdf
- [24] M. Marino, *The unexpected flight of the electron in a classical hydrogen-like atom*, J. Phys. A **36**, 11247 (2003), http://kirkmcd.princeton.edu/examples/EM/marino_jpa_36_11247_03.pdf